## Sweating the small stuff---the influence of metabolite extraction and separation on metabolomic studies

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## Resources

Metabolomics Workbench

www.metabolomicsworkbench.org

• XCMS Institute

 Great tutorials on chromatography, platforms, databases

• Twitter

## References

Mass spectrometry-based metabolomics

Xenobiotic metabolomics: major impact on the metabolome

The intestinal metabolome: an intersection between microbiota and host



"What have we here? A man or a fish? Dead or alive? He smells like a fish; a very ancient and fish-like smell . . ." W. Shakespeare The Tempest



GENOME FMO3 Mutation

FISHY EGGS

PROTEOME METABOLOME

# USE METABOLITE CHANGES TO INFORM ABOUT MECHANISM

## Metabolomics

- Metabolomics is the systematic analysis of the unique chemical fingerprints left behind by specific cellular processes
  - These small molecule metabolite profiles provide insight into cellular status.
- All "-omics" based scientific disciplines aim at the collective characterization and measurement of their particular constituent molecules
  - A comprehensive approach to study complete pools of biological molecules
  - Defines the structure, function and dynamics of an organism.
- Vast chemical diversity among small molecule metabolites has made extended coverage of the metabolome challenging
  - Size (50 1500 Da)
  - Concentration ( pM mM)
  - Physicochemical properties (diverse log P values)
  - Stereochemistry (distinct biological activity)

## Metabolite Extraction

- Currently no analytical technique exists that is capable of *in-situ* measurement of all classes of cellular metabolites
- Metabolite extraction therefore becomes a crucial step in any type of metabolomics study
  - Critical to both targeted and global based profiling strategies.
- Optimized extraction methodology should fulfill several criteria:
  - Extract the largest number of metabolites
  - Unbiased and non-selective physical or chemical properties of a molecule
  - Non-destructive no modification of metabolites

## Separation of Metabolites

- Mass spectrometry usually requires some form of chromatographic separation
  - Most systems use either liquid or gas chromatography
  - CE-MS gaining popularity
- Fractionation of sample components simplifies the resulting mass spectra while ensuring more accurate compound identification
  - Capacity factor (k) is critical to optimizing resolution
  - Increased resolution allows longer MS dwell times resulting in better signal/noise ratios
- Inadequate chromatographic separation of metabolites results in:
  - signal suppression ion suppression
  - compromised metabolite quantification
  - reduced metabolite coverage

| <b>Metabolite Class</b><br>Bile Acids - general  | Separation Mode<br>RPLC | Stationar<br>Waters B  | Phenomenex Luna NH2 silica<br>will break down over time at | .01%)              |
|--|-------------------------|------------------------|--|--------------------|
| Bile Acids - MCA isomers   | RPIC                    | Waters B               | high pH thus reducing column life, reducing                | .01%)              |
| Bile<br>Waters Amide column used<br>for HILIC will rapidly and<br>dramatically clog with salt<br>over time unless flushed<br>extensively at the end of each<br>run with<br>methanol/isopropanol/water. |                         | Restek Rapto           | resolving power, and dirtying the source.                  |                    |
|  |                         | Waters B<br>Waters CSF |  | .01%)<br>1 (0.01%) |
|  |                         | Waters BE              |  | ate (pH=10.0)      |
|  |                         | Waters (               |  | e (pH=10.0)        |
| م  |                         | Waters BE              | Thermo Hypercarb is robust,                                | e (pH=10.0)        |
| Eicosanoids  | RPLC                    | Waters B               | provides excellent   | .01%)              |
| Keto-prostaglandins  | RPLC                    | Waters B               | separation, but is extremely sticky.                       | .01%)              |
| Nucleotides  | HILIC                   | Waters BE              |  | e (pH=9.0)         |
| Nucleotides  | HILIC                   | Waters BE              |  | te (pH=9.0)        |
|  |                         | Thermo - Hyp           | ercarb   |                    |
| Nucleotides - NAD(P)+/NAD(P)H  | RPLC                    | graphit                | e acetonitrile/water/ammoniun                              | n acetate          |

## Hypothesis

Extraction and separation of metabolites may influence metabolomic studies as much as the disease process being investigated

### Rationale

Developing optimized protocols for extraction efficiency and chromatographic resolution based on metabolite class and/or characteristics will dramatically improve accuracy and reproducibility of metabolomic data sets.





aqueous phase

## Definitions

- Isomer same chemical formula, different chemical structure
- Stereoisomer same chemical formula, same order/sequence of bonded atoms, different 3dimensional orientation
- Isobar same mass, but different chemical formula

#### Resolution of Bile Acid Metabolites by RPLC using Waters BEH C18



#### Resolution of Taurine Conjugated MCA Isomers by RPLC on WATERS BEH C18



#### Resolution of Taurine Conjugated MCA Isomers by RPLC on Restek Rapture Biphenyl



#### Resolution of Taurine Conjugated MCA Isomers by RPLC on Restek Ultra AQ C18



## Tauro-β-Muricholic Acid An Typical Isomer Example in Metabolomics



### TβMCA is an Farnesoid X Receptor Antagonist

Shp (FXR target gene) induction in hepatocytes



Sayin et al *Cell Metabolism* 2013

Li F et al *Nature Communications* 2013

## METABOLOMICS FOR UNDERSTANDING DRUG TOXICITY---ACETAMINOPHEN



Bernard Brodie 1908-1989

#### ACETAMINOPHEN-INDUCED HEPATIC NECROSIS. I. ROLE OF DRUG METABOLISM'

J. R. MITCHELL, D. J. JOLLOW, W. Z. POTTER,"" D. C. DAVIS," J. R. GILLETTE AND B. B. BRODIE

Laboratory of Chemical Pharmacology, National Heart and Luny Institute, National Institutor of Health, Betherda, Maryland

ACETAMINOPHEN-INDUCED HEPATIC NECROSIS. II. ROLE OF COVALENT BINDING IN VIVO'

D. J. JOLLOW, J. R. MITCHELL, W. Z. POTTER, "\* D. C. DAVIS, J. R. GILLETTE AND B. B. BRODIE

Laboratory of Chemical Pharmacology, National Heart and Lung Institute, National Institutes of Health, Betheoda, Maryland

ACETAMINOPHEN-INDUCED HEPATIC NECROSIS. J11. CYTOCHROME P-450-MEDIATED COVALENT BINDING IN VITRO'

W. Z. POTTER,"" D. C. DAVIS," J. R. MITCHELL, D. J. JOLLOW, J. R. GILLETTE AND B. B. BRODIE

Laboratory of Chemical Pharmacology, National Heart and Lung Institute, National Institutes of Health, Bethesda, Meryland

ACETAMINOPHEN-INDUCED HEPATIC NECROSIS. IV. PROTECTIVE ROLE OF GLUTATHIONE'

J. R. MITCHELL, D. J. JOLLOW, W. Z. POTTER, J. R. GILLETTE AND B. B. BRODIE

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#### APAP NAPQI Toxicity

N-acetyl-p-benzoquinone imine

### NORMAL MOUSE LIVER

### NECROTIC MOUSE LIVER (400 mg/kg APAP 6 HOURS)



- Contained in 100s of products
- One of the most common pharmaceuticals associated with accidental and intentional poisoning (>7 g per adult per day)
- APAP overdose serves as a model for drug-induced liver toxicity
- Excess NAPQI (with reduced glutathione levels) leads to oxidative damage and inflammation leading to hepatocellular death/necrosis



### Acetaminophen Metabolomics



# Mouse Urinary Proteins (MUPs)

Dilute equal volume of mouse urine with an equal volume of 50% methanol



## **Mouse Urinary Proteins**

• Dilute equal volume of mouse urine with an equal volume of 100% methanol



### APAP Metabolism Study #4 Score Scatter Plot

PCA model



### **APAP Metabolism Loading Scatter Plot**

PCA Model



#### X-Variable Trend Plot for L-Carnitine (m/z=162.114+)



#### X-Variable Trend Plot for Propionylcarnitine (m/z=218.14+)



#### X-Variable Trend Plot for Acetylcarnitine (m/z=204.124+)



#### X-Variable Trend Plot for Decanoylcarnitine (m/z=316.247+)



## **INFLUENCE OF EXTRACTION PROTOCOL – Carnitines and CoAs**

#### CONTROL







### Influence of pH on Metabolite Extraction from Mouse Liver



### Influence of pH on Metabolite Extraction from Mouse Liver



### Influence of pH on Metabolite Extraction from Mouse Liver


### Extraction Efficiency of L-Carnitine from Mouse Liver



# Matrix Effects and Extraction

#### LIVER 8·10<sup>6</sup> 2.0.10 n-butanol IPA pH 4 6 · 10 ⁵ 1.5.10 IPA pH 8 MeOH:Water pH 4 area eakarea MeOH:Water pH 8 4·10 1.0.10 peak 2.10 5.0.10 C 0 C 2 C 3 C3:1 C 3 - O H C 4 C 4 - O H C4:1

#### SERUM



#### CARNITINES CO – C5

# Matrix Effects and Extraction

#### LIVER

#### SERUM



CARNITINES C10 – C14



#### Resolution of Acyl Carnitine Standards by RPLC on Waters BEH C18



### Acylcarnitine Extraction in Acidified IPA



#### **Increasing Chain Length**



#### Resolution of Coenzyme A (CoA) Thioester Metabolites by RPLC using Waters BEH C18



| 131120_           | CC_acylCoA           | _240                                    |      |   |   |      |      |                   |   |       |       |       |       |       |       |          |       |       |           |       |       |       | 19: MRM of                  | 4 Channels                  | ES+           |
|-------------------|----------------------|---|------|---|---|------|------|-------------------|---|-------|-------|-------|-------|-------|-------|----------|-------|-------|-----------|-------|-------|-------|-----------------------------|-----------------------------|---------------|
| <sup>100</sup> ]  |                      |   |      |   |   |      |      |                   |   |       |       |       |       |       |       |          |       |       | 18.95     |       |       | 10    | 132.775 > 428.              | 2.0 2.0                     | 07e6          |
| *                 |                      |   |      |   |   |      |      |                   |   |       |       |       |       |       |       |          |       |       |           |       |       |       |                             |                             |               |
| 0                 |                      |   |      |   |   |      |      |                   |   |       |       |       |       |       |       |          |       |       | $\square$ |       |       |       |                             |                             |               |
| -0.00             | ) 1.00               | 2.00                                    | 3.00 | 4.00                                    | 5.00                                    | 6.00 | 7.00 | 8.00              | 9.00                                    | 10.00 | 11.00 | 12.00 | 13.00 | 14.00 | 15.00 | 16.00    | 17.00 | 18.00 | 19.00     | 20.00 | 21.00 | 22.00 | 23.00 24                    | .00 25.00                   | )<br>594      |
| 100-              |                      | _201                                    |      |   |   |      |      |                   |   |       |       |       |       |       |       |          |       | 1     | 18.66     |       |       | 1030  | 1.755 > 428.35              | 9 (Lineleoyl (              | COA           |
|                   |                      |   |      |   |   |      |      |                   |   |       |       |       |       |       |       |          |       |       | 1         |       |       |       |                             | 1.3                         | 35e <i>i</i>  |
| ~                 |                      |   |      |   |   |      |      |                   |   |       |       |       |       |       |       |          |       |       | ]         |       |       |       |                             |                             |               |
| -0.00             | ) 1.00               | 2.00                                    | 3.00 | 4.00                                    | 5.00                                    | 6.00 | 7.00 | 8.00              | 9.00                                    | 10.00 | 11.00 | 12.00 | 13.00 | 14.00 | 15.00 | 16.00    | 17.00 | 18.00 | 19.00     | 20.00 | 21.00 | 22.00 | 23.00 24                    | .00 25.00                   | 0             |
| 131120_           | CC_acylCoA           | _239                                    |      |   |   |      |      |                   |   |       |       |       |       |       |       |          |       | 40.47 |           |       |       | 079   | 16: MRM of                  | 4 Channels                  | ES+           |
| <sup>100</sup> ]  |                      |   |      |   |   |      |      |                   |   |       |       |       |       |       |       |          |       | 18.17 |           |       |       | 970   | .109 / 420.34               | 8.3                         | 31e6          |
| *                 |                      |   |      |   |   |      |      |                   |   |       |       |       |       |       |       |          |       |       |           |       |       |       |                             |                             |               |
| 0                 |                      |   |      |   | ,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,, |      |      |                   |   |       |       |       |       |       |       |          |       |       |           |       |       |       |                             |                             |               |
| -0.00<br>131120 ( | ) 1.00<br>CC acviCoA | 2.00                                    | 3.00 | 4.00                                    | 5.00                                    | 6.00 | 7.00 | 8.00              | 9.00                                    | 10.00 | 11.00 | 12.00 | 13.00 | 14.00 | 15.00 | 16.00    | 17.00 | 18.00 | 19.00     | 20.00 | 21.00 | 22.00 | 23.00 24                    | .00 25.00<br>4 Channels     | 0<br>ES+      |
| 100-              |                      |   |      |   |   |      |      |                   |   |       |       |       |       |       |       |          | 17.3  | 5     |           |       |       | 9     | 50.734 > 428.3              | 59 (Lauroyl (               | CoA)          |
| <u></u>           |                      |   |      |   |   |      |      |                   |   |       |       |       |       |       |       |          |       |       |           |       |       |       |                             | 1.0                         | 5007          |
| Î                 |                      |   |      |   |   |      |      |                   |   |       |       |       |       |       |       |          |       |       |           |       |       |       |                             |                             |               |
| -0.00             | ) 1.00               | 2.00                                    | 3.00 | 4.00                                    | 5.00                                    | 6.00 | 7.00 | <mark>8.00</mark> | 9.00                                    | 10.00 | 11.00 | 12.00 | 13.00 | 14.00 | 15.00 | 16.00    | 17.00 | 18.00 | 19.00     | 20.00 | 21.00 | 22.00 | 23.00 24                    | .00 25.00                   | 0             |
| 131120_           | CC_acylCoA           | _234                                    |      |   |   |      |      |                   |   |       |       |       |       |       | 1     | 5 63     |       |       |           |       |       | 922   | 13: MRM of<br>665 > 415.552 | 4 Channels<br>2 (Decanovi ( | ES+<br>CoA    |
| 100               |                      |   |      |   |   |      |      |                   |   |       |       |       |       |       |       | Ň        |       |       |           |       |       |       |                             | 9.8                         | 87e7          |
| *                 |                      |   |      |   |   |      |      |                   |   |       |       |       |       |       |       | 1        |       |       |           |       |       |       |                             |                             |               |
| 0                 |                      | ,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,, |      | ,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,, | T                                       |      |      |                   | ,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,, | 1     |       | 1.1.1 |       |       |       | <u> </u> |       |       |           |       |       |       |                             |                             |               |
| -0.00<br>_131120_ | CC_acylCoA           | 2.00                                    | 3.00 | 4.00                                    | 5.00                                    | 6.00 | 7.00 | 8.00              | 9.00                                    | 10.00 | 11.00 | 12.00 | 13.00 | 14.00 | 15.00 | 16.00    | 17.00 | 18.00 | 19.00     | 20.00 | 21.00 | 22.00 | 23.00 24<br>12: MRM of      | 4 Channels                  | J<br>ES+      |
| 1007              |                      |   |      |   |   |      |      |                   |   |       |       | 11.93 |       |       |       |          |       |       |           |       |       | 894   | .571 > 428.295              | 5 (Octanoyl C<br>5.         | CoA )<br>14e7 |
| ~                 |                      |   |      |   |   |      |      |                   |   |       |       | A     |       |       |       |          |       |       |           |       |       |       |                             |                             |               |
|                   |                      |   |      |   |   |      |      |                   |   |       |       | Л     |       |       |       |          |       |       |           |       |       |       |                             |                             |               |
| -0.00             | ) 1.00               | 2.00                                    | 3.00 | 4.00                                    | 5.00                                    | 6.00 | 7.00 | 8.00              | 9.00                                    | 10.00 | 11.00 | 12.00 | 13.00 | 14.00 | 15.00 | 16.00    | 17.00 | 18.00 | 19.00     | 20.00 | 21.00 | 22.00 | 23.00 24                    | .00 25.00                   | 0             |
| 131120_0          | CC_acylCoA           | _232                                    |      |   |   |      |      | 7.65              |   |       |       |       |       |       |       |          |       |       |           |       |       | 866   | 10: MRM of<br>574 > 428.36  | 4 Channels<br>8 (Hexanoyl ( | ES+<br>CoA    |
| 100               |                      |   |      |   |   |      |      | Å.                |   |       |       |       |       |       |       |          |       |       |           |       |       |       |                             | 1.9                         | 97e7          |
| *                 |                      |   |      |   |   |      |      |                   |   |       |       |       |       |       |       |          |       |       |           |       |       |       |                             |                             |               |
| 0                 | 1.00                 | 2.00                                    | 3.00 | 4.00                                    | 5.00                                    | 6.00 | 7.00 | /\<br>8.00        | 9.00                                    | 10.00 | 11.00 | 12.00 | 13.00 | 14.00 | 15.00 | 16.00    | 17.00 | 18.00 | 10.00     | 20.00 | 21.00 | 22.00 | 23.00 24                    | T                           | ime           |
| -0.00             | 1.00                 | 2.00                                    | 0.00 | 4.00                                    | 0.00                                    | 0.00 | 1.00 | 0.00              | 5.00                                    | 10.00 | 11.00 | 12.00 | 10.00 | 14.00 | 10.00 | 10.00    | 17.00 | 10.00 | 19.00     | 20.00 | 21.00 | 22.00 | 20.00 24                    | .00 20.00                   | J             |

| 1311:                          | 20_CC_acylCoA_241                                       |              |      |      |       |       |       |                |       |       | 11: MRM of                                     | 4 Channels ES-                                       |
|--------------------------------|---|--------------|------|------|-------|-------|-------|----------------|-------|-------|--|--|
| 100<br>∎ ಕ                     |   |              | 5.69 |      |       |       |       |                |       |       | 886.574 > 428.358 (P                           | henylacetyl CoA<br>1.05e                             |
| 0<br><br>1311:<br>100<br>%     | 4<br>0.00 2.00<br>20_CC_acylCoA_231                     | 4.00<br>8.52 | 6.00 | 8.00 | 10.00 | 12.00 | 14.00 | 16.00          | 18.00 | 20.00 | 22.00 24<br>6: MRM of<br>838.588 > 428.        | 4.00<br>4 Channels ES-<br>415 (Butyri CoA<br>1.60e   |
| 0<br>بار<br>1311:<br>100<br>چ  | 20_00 2.00<br>20_CC_acyICoA_235<br>3.3                  | 4.00         | 6.00 | 8.00 | 10.00 | 12.00 | 14.00 | 16.00          | 18.00 | 20.00 | 22.00 24<br>7: MRM of<br>838.588 > 428.415     | 4.00<br>4 Channels ES-<br>i (Isobutyryl CoA<br>1.65e |
| 0<br>با<br>1311:<br>100<br>ځ   | 20_00 2.00<br>20_CC_acylCoA_244<br>2.77                 | 4.00         | 6.00 | 8.00 | 10.00 | 12.00 | 14.00 | 16 <u>.</u> 00 | 18.00 | 20.00 | 22.00 24<br>5: MRM of<br>836.514 > 428.291 (   | 4.00<br>4 Channels ES-<br>(2-butenoyl CoA<br>1.29e   |
| 0<br>ا-<br>1311:<br>100<br>چ   | 0.p0 2.00<br>20_CC_acyiCoA_230<br>1.98                  | 4.00         | 6.00 | 8.00 | 10.00 | 12.00 | 14.00 | 16.00          | 18.00 | 20.00 | 22.00 24<br>4: MRM of<br>824.568 > 428.362     | 4.00<br>4 Channels ES+<br>2 (Propionyl CoA<br>1.13e  |
| 0<br>با<br>1311:<br>100<br>کار | 20_CC_acyICoA_245<br>1.84<br>1.63                       | 4.00         | 6.00 | 8.00 | 10.00 | 12.00 | 14.00 | 16.00          | 18.00 | 20.00 | 22.00 24<br>1: MRM of<br>688.454 > 348.428 (3- | 4.00<br>4 Channels ES-<br>dephospho CoA<br>1.12e     |
| 0<br>با<br>1311:<br>100<br>چ   | 1,/U\<br>0.p0 2.00<br>20_CC_acyICoA_229<br>1.53<br>1.25 | 4.00         | 6.00 | 8.00 | 10.00 | 12.00 | 14.00 | 16 <u>.</u> 00 | 18.00 | 20.00 | 22.00 24<br>3: MRM of<br>810.538 > 428.        | 4.00<br>4 Channels ES-<br>367 (Acetyl CoA<br>3.66e   |
| 0<br>با<br>1311:<br>100<br>لا  | 0.p0 2.00<br>20_CC_acyICoA_228<br>1.44<br>1.10          | 4.00         | 6.00 | 8.00 | 10.00 | 12.00 | 14.00 | 16.00          | 18.00 | 20.00 | 22.00 24<br>2: MRM of<br>768.498 > 428.3       | 4.00<br>4 Channels ES-<br>75 (Coenzyme A<br>2.04e    |
| 0<br>-1                        | ₽ <mark>₽</mark> / <u>·</u> / <u>·</u>                  | 4.00         | 6.00 | 8.00 | 10.00 | 12.00 | 14.00 | 16.00          | 18.00 | 20.00 | 22.00 24                                       | 1.00   |





### AcylCoa Extraction via Modified Bligh/Dyer



#### Increasing Chain Length

Chem Res Toxicol. 2009 Apr;22(4):699-707. doi: 10.1021/tx800464q.

Serum metabolomics reveals irreversible inhibition of fatty acid beta-oxidation through the suppression of PPARalpha activation as a contributing mechanism of acetaminophen-induced hepatotoxicity.

Chen C<sup>1</sup>, Krausz KW, Shah YM, Idle JR, Gonzalez FJ.

J Biol Chem. 2008 Feb 22;283(8):4543-59. Epub 2007 Dec 19.

Identification of novel toxicity-associated metabolites by metabolomics and mass isotopomer analysis of acetaminophen metabolism in wild-type and Cyp2e1-null mice.

Chen C<sup>1</sup>, Krausz KW, Idle JR, Gonzalez FJ.

Hepatology. 2012 Jul;56(1):281-90. doi: 10.1002/hep.25645. Epub 2012 Jun 6.

Peroxisome proliferator-activated receptor alpha induction of uncoupling protein 2 protects against acetaminophen-induced liver toxicity.

Patterson AD<sup>1</sup>, Shah YM, Matsubara T, Krausz KW, Gonzalez FJ.



#### MITOCHONDRIAL DYSFUNCTION





# INFLUENCE OF EXTRACTION PROTOCOL – Ceramides

# Ceramide Physicochemical Properties

- Ceramides are a family of waxy lipid molecules.
  - Name derived from the latin word: cera = waxy + amide
- Ceramides are comprised of:
  - sphingosine: 18 carbon unsaturated amino alcohol
  - fatty acid moiety amide bond
- Ceramides are not water soluble:
  - Very hydrophobic
  - Confined to cellular membranes
  - Participate in lipid raft formation
  - >200 structurally distinct species have been identified in mammalian cells

### **Ceramide General Structure**



#### • Ceramide (d18:1/16:0)

- 2-amino-1,3-octadec-4-ene-diol
  - Amino alcohol (sphingoid) backbone
- Palmitic acid
  - Fatty acyl group

- Ceramide (d18:1/24:1(15Z))
- 2-amino-1,3-octadec-4-ene-diol
  - Amino alcohol (sphingoid) backbone
- 15-tetracosenoic acid
  - Fatty acyl group

# **Ceramide Biochemistry**

#### • Ceramides are found in high concentration in the membrane of cells.

- Structural component of the lipid bilayer
- Bioactive lipid implicated in a variety of physiological functions including:
  - ✓ Apoptosis and cell growth arrest
  - ✓ differentiation and cell senescence
  - ✓ cell migration and adhesion

#### • Ceramides are converted rapidly to more complex sphingolipids:

- Sphingomyelin
- Glycosylceramides
- Little accumulation observed
- Except for the skin (50% of total lipids can be ceramides)

# **Biosynthesis of Ceramides**

#### 1. De novo biosynthesis:

- Ceramide synthases couple sphinganine + long chain fatty acid to form dihydroceramide. •
- Double bond introduced into position 4 of the sphingoid base •
  - ceramide synthases 5 and 6 generate are specific for palmitic acid
  - ceramide synthases 1 (brain and skeletal muscle) specific for stearic aid (C18:0 & C18:1 ceramides)
  - ceramide synthases 2 specific for very long chain CoA-thioesters (C<sub>20</sub>-C<sub>26</sub>) (C20:0, C22:0, C24:0, C24:1 etc)
  - ceramide synthases 3 unusual ceramides of skin & testes
- 2. Catabolism of complex sphingolipids:
  - Sphingomyelinases/phospholipase C breakdown sphingomyelin in animal tissues •
  - Many factors can stimulate the hydrolysis of sphingomyelin to produce ceramide: •
    - Cytokines :TNF- $\alpha$ , IFN- $\gamma$  & various interleukins
    - 1,25-dihydroxy-vitamin  $D_3$
    - endotoxin
    - nerve growth factor
    - ionizing radiation & heat

- C16 ceramide

# **Ceramides and Disease**

- Ceramide metabolites have been implicated in various pathological conditions including:
  - Cancer
  - Diabetes
  - Obesity
  - Inflammation
  - Neurodegeneration
- Although not understood, the structure of individual ceramides aids in defining their physiological function.
  - Ceramides containing specific fatty acids are generated in response to particular stimuli.

# LC Method Development: Where to Start?

- Designing and optimizing an LC method involves choosing appropriate:
  - 1. Separation mechanism: NPC, RPLC, HILIC, size exclusion ion, exchange etc
  - 2. Column chemistry: C2, C4, C8, C18, cyanopropyl, phenyl, biphenyl, amide, SiOH etc
  - 3. Column properties: pore size, particle size & column dimensions
  - 4. Stationary and mobile phase combinations
- Critical to optimizing the chromatographic efficiency, retention, resolution & selectivity of analytes.

# Ceramide Scouting Gradients on Waters BEH C18



#### Fractionation of Ceramide Metabolites on Waters CSH C18 Column





#### Fractionation of Ceramide Metabolites on Waters CSH C18 Column



### **Ceramide Fragmentation Patterns**

*m/z* = 262 or 284









#### Ceramide Analysis in $CHCI_3$ : MeOH Extracted Murine Liver



# Conclusions

- Ceramides can be effectively resolved using reverse-phase liquid chromatography (RPLC) methodologies:
  - C18 column chemistry sufficient; but particle properties important (BEH vs CSH)
  - Stronger eluotropic series needed; MeOH or MeCN & water no good
  - Higher column temp required to compensate for increased backpressure (solvent viscosity & increased flow rates)
- Various ceramide metabolites can be detected using multiple UPLC-MS platforms:
  - Global metabolite profiling approach UPLC-ESI-QTOF-MS
  - Targeted metabolite approach UPLC-ESI-MS-MRM
- Poor detection of ceramides from bio-fluids and serum.
  - Low levels of endogenous ceramides?
  - Rapidly converted to more complex sphingolipids?
  - Ineffective extraction method?
  - Matrix effects Ion suppression?

# Conclusions

- Extraction protocols can impact metabolomic data sets considerably
- Solvent system composition and pH exhibit the most dramatic effects on metabolite recovery
  - The magnitude of these effects depend on metabolite class
  - Some classes of metabolites
- The number of extraction repetitions also plays a role in enhancing metabolite recovery
  - Tradeoff longer sample prep time
  - Larger sample volumes to process (evaporate)

# Conclusions

- Traditional RPLC methods can provide efficient separation of acyl-carnitine, bile acid and CoA thioester mixtures.
  - Advancements in hybrid particle technologies
  - Allowing for extremes in mobile phase pH and temperature manipulate selectivity
  - Complex ligand stationary phase interactions
- HILIC methods are superior at separating highly polar metabolites.
  - Nucleotides and derivatives
  - Small polar metabolites sugars, organic acids, amino acids, hydrophilic vitamins
- Advanced column chemistries (amide, aminopropyl, biphenyl, graphite, phenyl-hexyl) and alternative chromatographic methodologies (HILIC) can provide enhanced coverage of the metabolome.

### **Future Plans**

- There's no one "perfect" extraction or LC method available capable of efficiently resolving all components or features in the metabolome
- Therefore, our goal is to continue to develop optimized extraction and chromatography protocols for various classes of liver metabolites
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